Name: _____ Period: ____ Date: _____

GASES NOTES ADVANCED CHEMISTRY

Directions: This packet will serve as your notes for this chapter. Follow along with the PowerPoint presentation and fill in the missing information. Important terms / ideas are in all capitals and bolded!

•	Characteristics of Gases				
	1) Gases consist of hard,	ses consist of hard, particles (atoms or molecules)			
	2) Particles are	and do have			
	3) Large amounts of		_		
	4) Easily	and			
	5) Move	and			
•	Kinetic Theory of Gases				
	-Explain the	of gases and v	vhy they	the way they do	
	CONSTANT COMPRESSIBILIT		AND E) WITH		
•	Gas Laws				
	-Factors including will affect o			, and	
	-How a gas will to these changes can be predicted using certain				

	BOYLE'S LAW	CHARLES' LAW	GAY-LUSSAC'S LAW
Relationship			
Example			